

Monoclinic

*P*2<sub>1</sub>/c*a* = 8.6086 (3) Å*b* = 11.5661 (4) Å*c* = 14.7650 (5) Å $\beta$  = 91.554 (1) $^\circ$ *V* = 1469.58 (9) Å<sup>3</sup>*Z* = 4*D*<sub>x</sub> = 1.246 Mg m<sup>-3</sup>*D*<sub>m</sub> not measured**Data collection**

Siemens SMART CCD area-detector diffractometer

 $\omega$  scans

Absorption correction:

empirical (SADABS;

Sheldrick, 1996)

*T*<sub>min</sub> = 0.876, *T*<sub>max</sub> = 0.953

11 387 measured reflections

3622 independent reflections

**Refinement**Refinement on *F*<sup>2</sup>*R*[*F*<sup>2</sup> > 2*σ*(*F*<sup>2</sup>)] = 0.062*wR*(*F*<sup>2</sup>) = 0.125*S* = 1.000

3622 reflections

221 parameters

All H-atom parameters refined

Cell parameters from 3243 reflections  
 $\theta$  = 1.12–28.35°  
 $\mu$  = 0.487 mm<sup>-1</sup>  
*T* = 293 (2) K  
 Block  
 0.28 × 0.12 × 0.10 mm  
 Colourless

1757 reflections with  
 $I > 2\sigma(I)$   
 $R_{\text{int}}$  = 0.082  
 $\theta_{\text{max}} = 28.32^\circ$   
 $h = -11 \rightarrow 11$   
 $k = 0 \rightarrow 15$   
 $l = 0 \rightarrow 19$

$w = 1/[\sigma^2(F_o^2) + (0.042P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.278 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.385 \text{ e } \text{\AA}^{-3}$   
 Extinction correction: none  
 Scattering factors from  
*International Tables for Crystallography* (Vol. C)

**Table 1.** Selected geometric parameters (Å, °)

P1—N1	1.629 (2)	P1—Cl1	2.071 (1)
P1—C1	1.802 (3)	N1—C10	1.492 (4)
P1—S1	1.923 (1)	N1—C7	1.505 (4)
N1—P1—C1	105.1 (1)	N1—P1—Cl1	106.5 (1)
N1—P1—S1	117.9 (1)	C1—P1—Cl1	102.5 (1)
C1—P1—S1	113.4 (1)	S1—P1—Cl1	110.2 (1)

**Table 2.** Hydrogen-bonding geometry (Å, °)*C*<sub>g</sub> is the centroid of the C1–C6 ring.

<i>D</i> —H··· <i>A</i>	<i>D</i> —H	H··· <i>A</i>	<i>D</i> —··· <i>A</i>	<i>D</i> —H··· <i>A</i>
C2—H2···Cl1	0.85 (2)	2.70 (3)	3.180 (4)	117 (2)
C11—H11C··· <i>Cg1</i> <sup>1</sup>	0.90 (4)	2.93	3.577	130
C12—H12C··· <i>Cg1</i> <sup>1</sup>	0.93 (4)	3.03	3.897	156

Symmetry codes: (i)  $x, \frac{1}{2} - y, z - \frac{1}{2}$ ; (ii)  $-x, 1 - y, -z$ .

Data collection: SMART (Siemens, 1996). Cell refinement: SAINT (Siemens, 1996). Data reduction: SAINT. Program(s) used to solve structure: SHELXTL (Sheldrick, 1997). Program(s) used to refine structure: SHELXTL. Molecular graphics: XP in SHELXTL. Software used to prepare material for publication: SHELXTL, PLATON (Spek, 1990) and PARST (Nardelli, 1995).

SR and ES thank Dr M. N. S. Rao and Janardhanan, Indian Institute of Technology, Chennai, for the synthesis and supply of the compound. HKF thanks the Malaysian Government and Universiti Sains Malaysia for research grant R&D No. 190-9609-2801. SSSR

thanks the Universiti Sains Malaysia for a Visiting Postdoctoral Fellowship.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: SK1290). Services for accessing these data are described at the back of the journal.

**References**

- Bullen, G. J. & Tucker, P. J. (1972). *J. Chem. Soc. Dalton Trans.* pp. 1651–1658.  
 Cruickshank, D. W. J. (1964). *Acta Cryst.* **17**, 671–672.  
 Emsley, J. & Hall, D. (1976). *The Chemistry of Phosphorus*, p. 500. New York: Harper & Row.  
 Ismail, R. (1975). German Patent 1 543 539; *Chem. Abstr.* (1975), **83**, 97416q.  
 Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.  
 Sheldrick, G. M. (1996). SADABS. *Program for Empirical Absorption Correction of Area Detector Data*. University of Göttingen, Germany.  
 Sheldrick, G. M. (1997). SHELXTL. Version 5.1. *Software Reference Manual*. Bruker AXS Inc., Madison, Wisconsin, USA.  
 Siemens (1996). SMART and SAINT. *Area Detector Control and Integration Software*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.  
 Spek, A. L. (1990). *Acta Cryst. A* **46**, C-34.

*Acta Cryst.* (1999). **C55**, 1526–1528**9-(4H-1,2,4-Triazol-4-ylimino)-4,5-diazafluorene**

S. SHANMUGA SUNDARA RAJ,<sup>a</sup> HOONG-KUN FUN,<sup>a</sup> DUN-RU ZHU,<sup>b</sup> FANG-FANG JIAN,<sup>b</sup> KOULIN ZHANG<sup>b</sup> AND XIAO-ZENG YOUNG<sup>b</sup>

<sup>a</sup>X-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia, and

<sup>b</sup>Coordination Chemistry Institute & State Key Laboratory of Coordination Chemistry, Nanjing University, Nanjing 210093, People's Republic of China. E-mail: hkfun@usm.my

(Received 12 May 1999; accepted 8 June 1999)

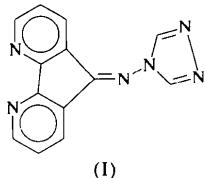
**Abstract**

The title molecule, C<sub>13</sub>H<sub>8</sub>N<sub>6</sub>, is non-planar. The dihedral angle between the planes of the diazafluorene moiety and the triazole ring is 61.0 (1) $^\circ$ . The molecules are arranged as chains running along the *c* axis and along the normal to the *bc* plane through C—H···N hydrogen bonds.

**Comment**

The aroyl Schiff bases of 4,5-diazafluoren-9-one have received considerable attention over the past few decades

(Wang & Rillema, 1997; Duan *et al.*, 1992). These compounds may function as polypyridyl-bridged ligands (Wang *et al.*, 1998). Schiff base compounds having extensively conjugated  $\pi$ -electron systems have been reported to display second harmonic generation (Houlton *et al.*, 1992), however, the heteroaroyl Schiff bases have rarely been studied. As a continuation of our investigation of the structure of 4,5-diazafluoren-9-one (Fun *et al.*, 1995; Lu *et al.*, 1995, 1996), we have determined the crystal structure of 9-(4*H*-1,2,4-triazol-4-ylimino)-4,5-diazafluorene, (I).



(I)

The X-ray analysis of (I) indicated that it is non-planar (Fig. 1). The bond lengths and angles observed in the structure are normal. The dihedral angle between the planes of the diazafluorene moiety and the triazole ring is  $61.0(1)^\circ$ , whereas that between the diazafluorene moiety and the pyridine ring in 4,5-diaza-9-fluorene isonicotinoyl hydrazone monohydrate (Lu *et al.*, 1996) is  $27.74(4)^\circ$ . The interplanar angle between rings A and B is  $0.7(1)^\circ$ , but that between rings B and C is  $2.6(1)^\circ$ . This increase in the dihedral angle may be due to steric repulsion between C2 and C12 [C2 $\cdots$ C12 3.166(3) Å].

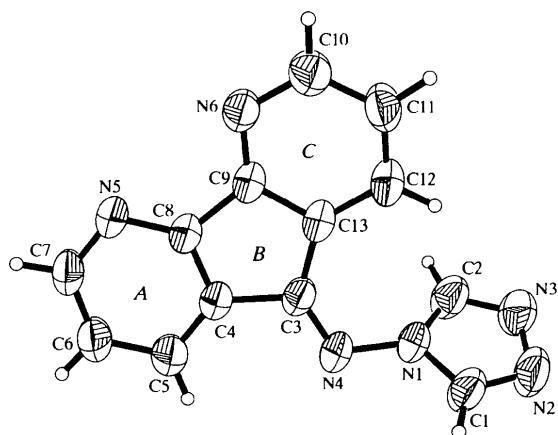


Fig. 1. The structure of title compound showing 50% probability displacement ellipsoids and the atom-numbering scheme.

We have observed previously that due to the possibility of different charge concentrations between the two pyridyl N atoms of the diazafluorene moiety, the one with greater charge is involved in the stronger hydrogen bonds, whereas the other may be involved in a weak interaction (Fun *et al.*, 1995; Lu *et al.*, 1995). It can be

seen from Fig. 2 that there are C—H $\cdots$ N hydrogen-bonded chains along the normal to the *bc* plane and chains along the *c* axis.

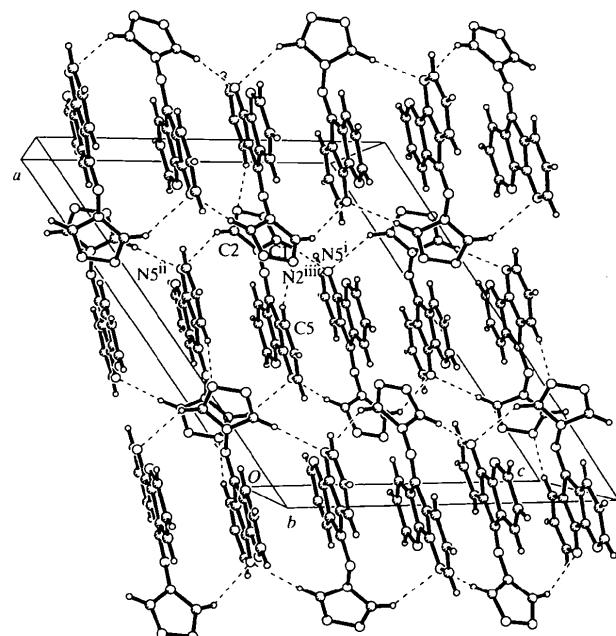


Fig. 2. The packing of molecules of (I) viewed down the *b* axis showing the C—H $\cdots$ N hydrogen bonds

The molecules are stacked along the *c* axis through interactions between ring centroids (*Cg*). The distances between centroids are *Cg*1 $\cdots$ *Cg*1( $1-x, y, \frac{1}{2}-z$ ) 3.674(4) Å and *Cg*1 $\cdots$ *Cg*2( $1-x, 1-y, -z$ ) 3.512(5) Å, where *Cg*1 and *Cg*2 are the centroids of rings B and C, respectively.

## Experimental

The title compound was prepared by condensation of 4,5-diazafluoren-9-one and 4-amino-4*H*-1,2,4-triazole in ethanol. X-ray quality crystals were obtained by recrystallization from an *N,N*-dimethylformamide–ether solution at room temperature.

### Crystal data

$C_{13}H_{8}N_6$	Mo $K\alpha$ radiation
$M_r = 248.25$	$\lambda = 0.71073$ Å
Monoclinic	Cell parameters from 2725 reflections
$C2/c$	$\theta = 2.36\text{--}28.32^\circ$
$a = 18.304(1)$ Å	$\mu = 0.096$ mm $^{-1}$
$b = 10.488(1)$ Å	$T = 293(2)$ K
$c = 14.267(1)$ Å	Block
$\beta = 123.790(1)^\circ$	$0.36 \times 0.16 \times 0.14$ mm
$V = 2276.2(3)$ Å $^3$	Yellow
$Z = 8$	
$D_x = 1.449$ Mg m $^{-3}$	
$D_m$ not measured	

*Data collection*

Siemens SMART CCD area-detector diffractometer  
 $\omega$  scans  
 Absorption correction: none  
 7790 measured reflections  
 2817 independent reflections

1553 reflections with  
 $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.054$   
 $\theta_{\text{max}} = 28.29^\circ$   
 $h = -24 \rightarrow 22$   
 $k = -13 \rightarrow 12$   
 $l = -18 \rightarrow 19$

*Refinement*

Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.063$   
 $wR(F^2) = 0.197$   
 $S = 1.018$   
 2817 reflections  
 172 parameters  
 H atoms constrained  
 $w = 1/[\sigma^2(F_o^2) + (0.1040P)^2]$   
 where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.197 \text{ e } \text{\AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.228 \text{ e } \text{\AA}^{-3}$   
 Extinction correction: none  
 Scattering factors from  
*International Tables for Crystallography* (Vol. C)

Fun, H.-K., Sivakumar, K., Zhu, D.-R. & You, X.-Z. (1995). *Acta Cryst.* **C51**, 2076–2078.

Houlton, A., Jasim, N., Roberts, R. M. G., Silver, J., Cunningham, D., McArdle, P. & Higgins, T. (1992). *J. Chem. Soc. Dalton Trans.* pp. 2235–2241.

Lu, Z.-L., Duan, C.-Y., Tian, Y.-P., You, X.-Z., Fun, H.-K. & Sivakumar, K. (1995). *Acta Cryst.* **C51**, 2078–2080.

Lu, Z.-L., Shan, B.-Z., Duan, C.-Y., Tian, Y.-P., You, X.-Z., Fun, H.-K. & Sivakumar, K. (1996). *Acta Cryst.* **C52**, 1720–1722.

Nardelli, M. (1995). *J. Appl. Cryst.* **28**, 659.

Sheldrick, G. M. (1997). *SHELXTL Software Reference Manual*.

Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.

Siemens (1996). *SMART and SAINT. Area Detector Control and Integration Software*. Siemens Analytical X-ray Instruments Inc., Madison, Wisconsin, USA.

Spek, A. L. (1990). *Acta Cryst.* **A46**, C-34.

Wang, Y., Perez, W. J., Zheng, G. Y., Rillema, D. P. & Huber, C. L. (1998). *Inorg. Chem.* **37**, 2227–2234.

Wang, Y. & Rillema, D. P. (1997). *Tetrahedron*, **53**, 12377–12390.

Table 1. Selected geometric parameters ( $\text{\AA}$ ,  $^\circ$ )

N1—C1	1.341 (3)	N4—C3	1.280 (3)
N1—C2	1.353 (3)	N5—C8	1.336 (3)
N1—N4	1.410 (3)	N5—C7	1.337 (3)
N2—C1	1.299 (3)	N6—C9	1.321 (3)
N2—N3	1.381 (3)	N6—C10	1.349 (3)
N3—C2	1.289 (3)		
C1—N1—C2	104.5 (2)	C3—N4—N1	116.0 (2)
C1—N1—N4	124.9 (2)	N4—C3—C13	134.6 (2)
C2—N1—N4	129.5 (2)		

Table 2. Hydrogen-bonding geometry ( $\text{\AA}$ ,  $^\circ$ )

$D—H \cdots A$	$D—H$	$H \cdots A$	$D \cdots A$	$D—H \cdots A$
C1—H1A $\cdots$ N5 <sup>i</sup>	0.93	2.57	3.381 (4)	145
C2—H2A $\cdots$ N5 <sup>ii</sup>	0.93	2.50	3.361 (4)	153
C5—H5A $\cdots$ N2 <sup>iii</sup>	0.93	2.62	3.513 (4)	162

Symmetry codes: (i)  $1 - x, y, \frac{1}{2} - z$ ; (ii)  $1 - x, 1 - y, -z$ ; (iii)  $\frac{1}{2} - x, \frac{1}{2} - y, -z$ .

Data collection: *SMART* (Siemens, 1996). Cell refinement: *SAINT* (Siemens, 1996). Data reduction: *SAINT*. Program(s) used to solve structure: *SHELXTL* (Sheldrick, 1997). Program(s) used to refine structure: *SHELXTL*. Molecular graphics: *SHELXTL*. Software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 1990).

This work was supported financially by the State Science and Technology Commission and the National Nature Science Foundation of China. The authors would like to thank the Malaysian Government and Universiti Sains Malaysia for research grant R&D No. 190-9609-2801. SSSR thanks the Universiti Sains Malaysia for a Visiting Postdoctoral Fellowship.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: NA1425). Services for accessing these data are described at the back of the journal.

## References

Duan, C.-Y., Zhu, L.-G. & You, X.-Z. (1992). *Acta Chim. Sin.* **50**, 715–720.

*Acta Cryst.* (1999). **C55**, 1528–1530

## 8-Méthyl-1,3-diphényl-4,5,6,6a,7,8-hexa-hydrocyclopenta[b][1,2,4]triazolo[4,3-a]-benzodiazépin-7-one

A. AATIF,<sup>a</sup> A. BAOUID,<sup>a</sup> H. HASNAOUI,<sup>a</sup> A. BENHARREF<sup>a</sup> ET M. PIERROT<sup>b</sup>

<sup>a</sup>Laboratoire de Chimie des Substances Naturelles, Département de Chimie, Faculté des Sciences Semlalia, Université Cadi Ayyad, Marrakech, Maroc, et <sup>b</sup>LBS-UMR 6517, Centre Scientifique Saint-Jérôme, 13397 Marseille, CEDEX 20, France. E-mail: marcel.pierrot@lbs.u-3mrs.fr

(Received 1 mars 1999, accepted 17 mai 1999)

## Abstract

The structure of the title compound, C<sub>26</sub>H<sub>24</sub>N<sub>4</sub>O, shows that the dipolar cycloaddition reactions of nitrilamines with benzodiazepines are peri- and regioselective. The central seven-membered ring is fused with a triazolo ring, a five- and a six-membered ring to form the core of the molecule.

## Commentaire

La péri- et la régiosélectivité des réactions de cycloaddition dipolaires-1,3 des nitrilamines avec des 1,5- et 1,4-benzodiazépines ont déjà été mises en évidence à plusieurs reprises par notre équipe (Benelbaghdadi *et al.*, 1998; Essaber, Baoudi, Hasnaoui, Benharref & Lavergne, 1998; Baoudi *et al.*, 1996; Essaber, Baoudi, Hasnaoui, Giorgi & Pierrot, 1998; Chiaroni, Riche,